

# Selectively dispersed isotope labeling for protein structure determination by magic angle spinning NMR

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**Abstract** The power of nuclear magnetic resonance spectroscopy derives from its site-specific access to chemical, structural and dynamic information. However, the corresponding multiplicity of interactions can be difficult to tease apart. Complimentary approaches involve spectral editing on the one hand and selective isotope substitution on the other. Here we present a new “redox” approach to the latter: acetate is chosen as the sole carbon source for the extreme oxidation numbers of its two carbons. Consistent with conventional anabolic pathways for the amino acids, [1-<sup>13</sup>C] acetate does not label  $\alpha$  carbons, labels other aliphatic carbons and the aromatic carbons very selectively, and labels the carboxyl carbons heavily. The benefits of this labeling scheme are exemplified by magic angle spinning spectra of microcrystalline immunoglobulin binding protein G (GB1): the elimination of most J-couplings and one- and two-bond dipolar couplings provides narrow signals and long-range, intra- and inter-

residue, recoupling essential for distance constraints. Inverse redox labeling, from [2-<sup>13</sup>C] acetate, is also expected to be useful: although it retains one-bond couplings in the sidechains, the removal of CA–CO coupling in the backbone should improve the resolution of NCACX spectra.

**Keywords** Isotope substitution · Sparse labeling · MAS NMR structure determination · Dipolar truncation · Isotope labeled peptone · <sup>13</sup>C-acetate

## Introduction

The interactions between nuclear spins are important sources of information in nuclear magnetic resonance (NMR) spectroscopy. However, dense networks of nuclear spins can obscure that information via line broadening, spectral overlap, and truncation of polarization transfers between weakly coupled spins, especially in solid state NMR. These countervailing considerations have motivated and informed different isotope substitution schemes. In magic angle spinning (MAS) NMR, uniform <sup>13</sup>C and <sup>15</sup>N labeling has been widely used for resonance assignments and structure determination. At the same time, non-uniform labeling has shown great potential to extend these experiments to larger systems. Approaches have included the use of reverse (<sup>12</sup>C) labeling of specific amino acids (Shi et al. 2009; Etzkorn et al. 2007), specific (or ‘forward’) labeling of selected amino acids (Hiller et al. 2008), labeling from [2-<sup>13</sup>C] or [1,3-<sup>13</sup>C] glycerol (Hong 1999; LeMaster and Kushlan 1996; Sperling et al. 2010; Bayro et al. 2010; Higman et al. 2009), and labeling from [1-<sup>13</sup>C] or [2-<sup>13</sup>C] glucose (Lundstrom et al. 2007; Loquet et al. 2011). Glycerol labeling has proved useful in determining the

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structures of monomeric proteins (Nieuwkoop et al. 2009; Castellani et al. 2002, 2003), while the determination of multimeric protein structures has been facilitated by heterogeneous labeling via glycerol (Nieuwkoop and Rienstra 2010; Bayro et al. 2011; Debelouchina et al. 2010) or glucose (Loquet et al. 2010, 2012; Lv et al. 2012).

While current non-uniform labeling schemes have been useful for removing one-bond scalar and dipolar couplings, two-bond couplings remain and interfere with quantitative measurements of long-range (4–5 Å or greater) homonuclear  $^{13}\text{C}$  distances via dipolar truncation (Bayro et al. 2009; Costa 1996), in which more strongly coupled spins significantly attenuate polarization transfer among weakly coupled sites for most zero quantum (ZQ) and double quantum (DQ) recoupling methods. Thus, more dilute labeling, to further attenuate dipolar truncation, is required to improve recoupling transfer efficiencies and allow quantitative measurement of longer homonuclear distances. A previous study by McDermott and coworkers (Goldbourn et al. 2007) showed that  $[1-^{13}\text{C}]$  glucose in *Pseudomonas aeruginosa* K could selectively label carbonyl sites via the Entner-Doudoroff pathway. However, almost no sidechain sites were labeled in this scheme, limiting the application of broadband recoupling techniques to measure structurally important distance constraints.

Here we present a new non-uniform labeling scheme that uses acetate as the carbon source. With two carbons in highly divergent oxidation states, acetate entering the tricarboxylic acid cycle eventually deposits its carboxyl carbon in the carboxyl group of almost all amino acids and at single sidechain sites for a majority of amino acids. As a result of this pattern, which we call “redox labeling”, many of the remaining homonuclear dipolar couplings, including backbone-to-sidechain and sidechain-to-sidechain couplings, are of similar magnitude. This significantly reduces dipolar truncation compared to other non-uniform labeling schemes and permits long range contacts to be measured simultaneously in a broadband recoupling experiment.

Since few organisms can survive on acetate as the sole carbon source, our labeling protocol involves preparing a labeled peptone (i.e., whole cell hydrolysate) from an organism that can, and feeding that peptone to cells expressing the protein of interest. To prepare the peptone from acetate, we have tried both *Escherichia coli* NCM 533 and *Pseudomonas mendocina* Palleroni. Since the distribution of acetate label is more sharply defined in the *Pseudomonas* derived peptone, we present only that procedure. To illustrate the benefits of the labeling scheme in MAS NMR of proteins, we present spectra of the beta 1 domain from immunoglobulin binding protein G (GB1) expressed in *E. coli* BL21(DE3) cells grown on the *Pseudomonas* derived peptone.

## Materials and methods

### Preparation of redox-labeled peptone

Lyophilized *Pseudomonas mendocina* Palleroni (ATCC 25411) bacteria were revived in nutrient broth (Difco), and grown on “standard media” (Stanier et al. 1966) with sodium acetate as the sole carbon source (natural abundance control) or with acetate replaced by  $[1-^{13}\text{C}]$  acetate (Cambridge Isotope Labs, Andover MA). The bacteria were collected and lyophilized for weighing. For hydrolysis, the cells were resuspended in 6 M HCl, deaerated by bubbling with nitrogen or argon, and incubated at 70 °C for 3 days. Excess HCl was subsequently removed by heating under vacuum and adding water periodically to avoid drying.

### Preparation of redox-labeled amino acids for analysis of isotope incorporation

Collected *P. mendocina* were resuspended in pH 7.4, 20 mM Tris, 100 mM NaCl buffer and lysed by sonication. The lysate was centrifuged at 9,000g to separate cytosolic protein from other cell components. The cytosolic protein was precipitated from the supernatant by adding (slowly on ice) saturated ammonium sulfate up to 75 % (Jakoby 1971). The precipitated protein was collected by centrifugation at 9,000g for 15 min and dialyzed against pH 7.4 buffer, using SpectaPor7 membrane (MWCO 3500). The protein concentration was determined colorimetrically (Quick Start Bradford protein assay, BioRad). The suspension was mixed with an equal volume of 12 M HCl, deaerated by replacing air with argon in several freeze-thaw degassing cycles, and hydrolyzed overnight at 110 °C. After most of the HCl was removed by repeated lyophilization and resuspension in  $\text{H}_2\text{O}$  (or  $\text{D}_2\text{O}$ ), the pH was adjusted to neutral with NaOH (or NaOD).

### MS/MS analysis of redox-labeled amino acids

Arrays of LC–ESI–MS/MS spectra were run for both the labeled and control peptones, using a Finigan LCQ ion trap and Agilent HPLC. Reverse phase HPLC of the amino acids was conducted with the volatile ion pairing agent, perfluoroheptanoic acid (PFHA) added to the mobile phase to improve the separation (Petritis et al. 2000; Qu et al. 2002a, b). Chromatography was carried out over 35 min, at 40 °C, on a Luna C18 (Phenomenex) 4.6 mm × 150 mm column, with particle size 3 μm, an injection volume of 100 μl, and a flow rate of 1 ml/min. The two mobile phase solvents were (A) 0.8 mM PFHA and 0.005 % TFA in water, and (B) acetonitrile. The proportion of solvent A was maintained at 100 % for the first 6 min, reduced to

76 % over the following 8 min, maintained at 76 % for the next 7 min, decreased to 40 % over the following 6 min, and maintained at 40 % to the end of the run. The LC effluent was split 10:1 before the interface. The collision energy was maintained at 30 eV. All spectra were taken in positive ion mode.

#### NMR analysis of redox-labeled amino acids

One-dimensional  $^{13}\text{C}$  NMR spectra in  $\text{D}_2\text{O}$  were taken on a Varian Inova 400 MHz spectrometer.

Two-dimensional spectra were acquired at 600 MHz  $^1\text{H}$  Larmor frequency on a custom-built spectrometer (D. J. Ruben). Double quantum filtered COSY spectra were acquired with 8 kHz spectral width in each dimension, using 2,048 and 4,096 data points in the F1 and F2 dimensions, respectively. HSQC spectra were acquired with 12 and 6 kHz spectral widths in the F1 and F2 dimensions, respectively, using 1,024 data points in each dimension. The spectra were inspected using Sparky (T. D. Goddard and D. G. Kneller and University of California), and  $^1\text{H}$  chemical shifts were compared with  $^{13}\text{C}$  chemical shifts from HSQC, leading to assignment of amino acid type. Intensities of cross peaks and their satellites were obtained by line fitting, and estimates of the degree of  $^{13}\text{C}$  labeling were calculated as the ratio of the sum of the satellites to the cross peak intensity for the  $^1\text{H}$  bound to the site of  $^{13}\text{C}$  labeling. COSY cross peaks with intense satellites were in general correlated with very intense HSQC signals.

#### Preparation of GB1 samples

The T2Q construct of the beta 1 domain from immunoglobulin binding protein G (GB1) was transformed into *E. coli* BL21(DE3) competent cells (Invitrogen Corp.). These cells were grown in M9 media with 1 g  $^{15}\text{N}$  ammonium chloride (Cambridge Isotope Labs) per liter as the sole nitrogen source, and either redox-labeled peptone prepared from 4 g of dried *Pseudomonas* or 3 g of  $[\text{U}-^{13}\text{C}]$  glucose (Cambridge Isotope Labs) per liter as the sole carbon source, to obtain  $[\text{redox-}^{13}\text{C}, \text{U}-^{15}\text{N}]$  GB1 or  $[\text{U}-^{13}\text{C}, ^{15}\text{N}]$  GB1, respectively. The doubling times in the media containing peptone were very similar to doubling times in Luria broth, as expected for complex media.

Microcrystalline samples of GB1 were prepared according to previously published protocols (Franks et al. 2005; Schmidt et al. 2007). The soluble fraction released from cells by sonication was heated at 80 °C for 5 min, cooled on ice for 15 min, and then separated on a size exclusion column. Fractions containing GB1 were pooled and extensively dialyzed against 50 mM sodium phosphate, pH 5.5, prior to forming microcrystals. To reduce intermolecular contacts, redox-labeled GB1 was diluted with natural abundance GB1

at a molar ratio of 1:3 and spectra were obtained from a total of ~40 mg of GB1, containing 10 mg of redox-labeled GB1. In the case of  $[\text{U}-^{13}\text{C}, ^{15}\text{N}]$  GB1, spectra were obtained from a ~25 mg sample.

#### MAS NMR studies of GB1

All MAS NMR data were acquired on a Bruker 800 MHz Avance III spectrometer running Topspin 3.1. Each sample was centrifuged into a separate Bruker 3.2 mm rotor, and the drive tips were sealed with epoxy to ensure that the samples remained hydrated.

2D  $^{13}\text{C}$ - $^{13}\text{C}$  RFDR (Bennett et al. 1998) correlation experiments were acquired at 18.0 kHz MAS frequency. RFDR transfers employed 83 kHz  $^{13}\text{C}$   $\pi$ -pulses and 105 kHz  $^1\text{H}$  CW decoupling with the  $^{13}\text{C}$  RF carrier set to 100 ppm. RFDR transfers were optimized in 1D experiments by varying the decoupling field strength from 90 to 110 kHz in order to minimize interference with the  $^1\text{H}$  decoupling (Bayro et al. 2008). 83 kHz TPPM (Bennett et al. 1995)  $^1\text{H}$  decoupling was employed during the evolution and acquisition periods, with a 5.8  $\mu\text{s}$  pulse length and relative phases of 0° and 20°. 2,048 points were acquired in t2 and  $512 \times 18 \mu\text{s}$  (total evolution time 20 ms) were acquired in t1, with 8 scans per point and a recycle delay of 2 s, for a total acquisition time of approximately 10 h per 2D experiment.

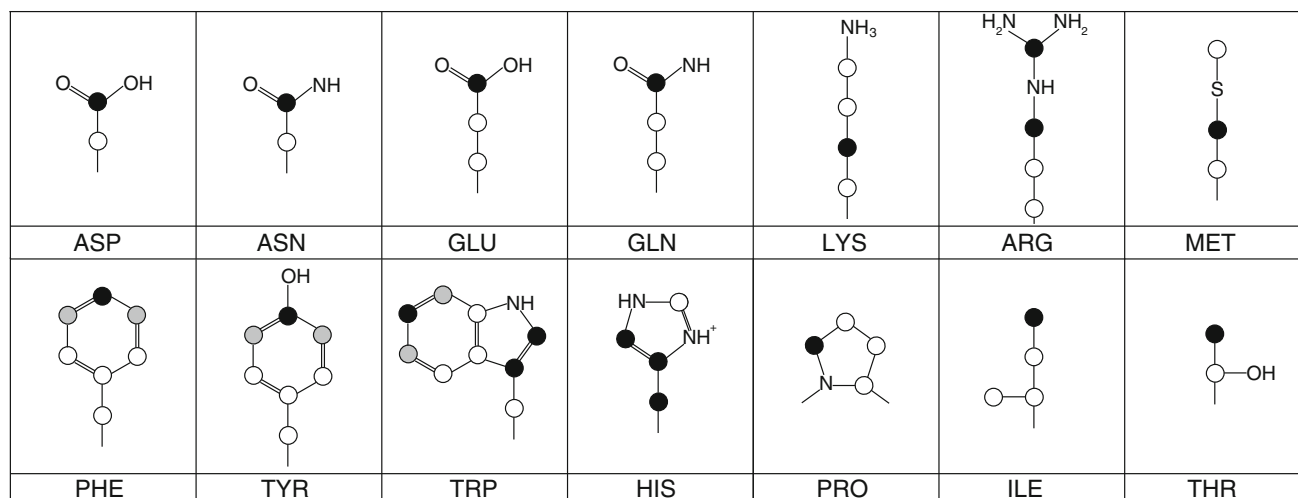
2D  $^{15}\text{N}$ - $^{13}\text{C}$  ZF-TEDOR (Jaroniec et al. 2002; Hing et al. 1992) experiments were acquired at 18.0 kHz MAS frequency. During the TEDOR recoupling period, 45 kHz  $^{15}\text{N}$   $\pi$ -pulses were employed with 83 kHz CW  $^1\text{H}$  decoupling. 83 kHz TPPM  $^1\text{H}$  decoupling was also employed during the evolution and acquisition periods, with 5.8  $\mu\text{s}$  pulse length and relative phases of 0° and 20°.

nmrPipe (Delaglio et al. 1995) and Sparky (T. D. Goddard and D. G. Kneller and University of California) were used to process data and generate figures, respectively. The same processing and display parameters were used for comparing spectra between redox and uniformly labeled GB1. Processing details are given in each figure caption.

## Results

#### Peptone analysis

Figure 1 shows the pattern of sidechain labeling from  $[\text{U}-^{13}\text{C}]$  acetate that is expected from established anabolic pathways. In addition, the backbone carbonyl is expected to be labeled in all the amino acids except histidine. Contiguous labeling is expected only in the aromatics that incorporate erythrose and ribose, as these sugars are serially labeled in the pentose phosphate pathway (from



**Fig. 1** Diagram of the expected side chain distribution of  $^{13}\text{C}$  from  $[1-^{13}\text{C}]$  acetate. *Black circles* identify heavily labeled sites and *gray circles* identify sites that are labeled more lightly because symmetric

carbons of 3-dehydroquinate can arise equally from labeled and unlabeled carbons of erythrose 4-phosphate in the shikimate pathway

glyceraldehyde 3-phosphate and fructose 6-phosphate labeled in gluconeogenesis).

Comparison of the 1D  $^{13}\text{C}$  NMR spectrum of redox-labeled amino acids with that of the natural abundance control showed the expected labeling pattern: the  $\alpha$  carbons (42 ppm for glycine and 51–61 ppm otherwise) are not labeled, the other aliphatic carbons (12–41 ppm) and the aromatic carbons (108–156 ppm) are selectively labeled, and the carboxyl carbons (shifted to 166–178 ppm due to the low pH of the solution) are heavily labeled.

As shown in Table 1, the MS and MS/MS data report the number of carbon atoms labeled in individual amino acids, and for some amino acids the exact location of the label. We were able to separate and analyze most amino acids. Analysis of glycine was not possible due to very high solvent background in the vicinity of 80  $m/z$ . This problem was also reported by Petritis et al. (2000). We were also unable to obtain MS/MS data for cysteine, tryptophan and histidine due to small MS peaks. Tryptophan is partially destroyed during acid hydrolysis.

For most of the amino acids, the most common product of collisionally activated dissociation was  $(M + H - 46)^+$ , which corresponds to the neutral loss of a formic acid by a rearrangement (Qu et al. 2002b). In each case, the data in Table 1 show that the remaining ( $-\text{HCOOH}$ ) fragment has a lower distribution of masses than the intact amino acid. This indicates that the carboxyl group is labeled in most of the amino acids.

For some amino acids, only one site is labeled (mass increment = +1) and that is the carboxyl group. For example, the MS spectrum of valine (Fig. 2) shows peaks at  $m/z$  118.0 and 119.0 (vs. 118 alone in the control). The relative intensities suggest that a little more than half of the

valines are singly labeled, and the rest are unlabeled. Since the mass of the  $-\text{HCOOH}$  fragment in the MS/MS spectrum is the same for the labeled valine and the control ( $m/z = 72$ ), only the cleaved carboxyl group was labeled. Similarly, alanine, leucine and serine are labeled only in the carboxyl group. Whereas the labeling of the carboxyl is nearly complete in leucine, the comparative peak intensities for alanine ( $m/z = 91$ ,  $m/z = 90$ ) indicate that the labeling there is about two thirds.

Other amino acids are more extensively labeled. Aspartic acid is labeled in up to two sites (although mostly singly labeled) and MS/MS shows that one or two carboxyl groups are labeled. Glutamic acid labeling is very similar to aspartic acid, except that most of the residues are doubly labeled. Isoleucine, lysine, methionine, proline and threonine are also labeled in up to two sites (including the carboxyl). For the protonated carbons, analysis of COSY intensities (also shown in Table 1) indicates 62 % labeling of CD1 in isoleucine, 60 % labeling of CG in lysine, 63 % labeling of CG in methionine,  $\sim 99$  % labeling of CD in proline, and 67 % labeling of CG in threonine.

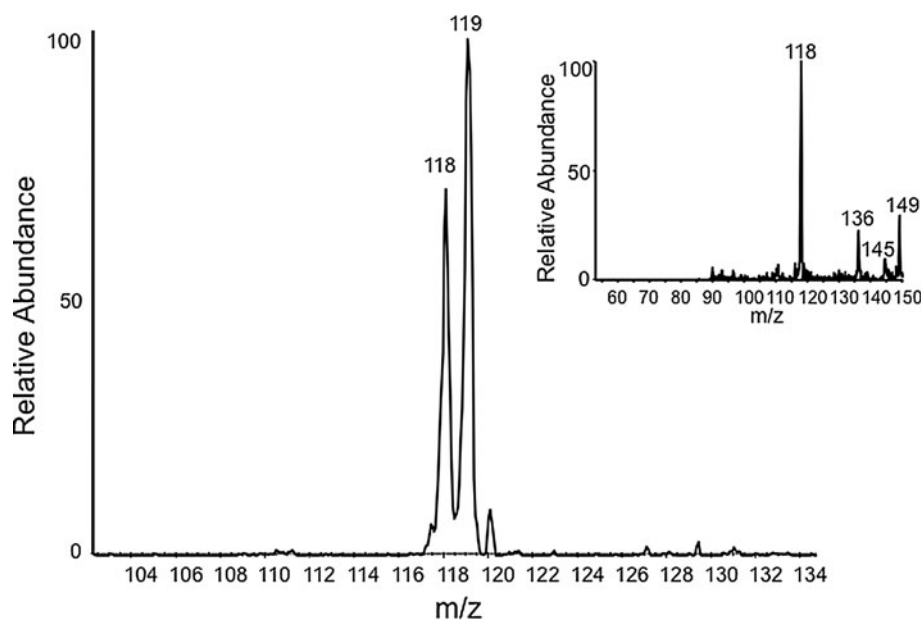
Arginine, phenylalanine and tyrosine are labeled at up to three sites, including the carboxyl. In arginine, the fragment masses show that the guanido group (CZ) is labeled and COSY shows that CD is 80 % labeled. In phenylalanine, the MS3 fragment resulting from removal of the carboxyl group and the benzyl group ( $m/z = 44$ , not shown in the table) has no label, indicating that the ring has up to two labeled carbons. In histidine and tryptophan, the number of sites labeled was up to two and four, respectively. These are the only residues for which the observed extent of labeling is less than expected.

**Table 1** Analysis of [redox-<sup>13</sup>C] labeled peptone

Amino acid (MS) /fragment (MS2)	Control (M/Z)	Labeled mass increment					Labels expected	COSY result
		+0	+1	+2	+3	+4		
Ala	90	+	++				C	
–HCOOH	44	++					–	
Arg	175		+	++	+		C, CD, CZ	
–NH3	158		+	++	+		C, CD, CZ	
–COOH	130		++	+			CD, CZ	
–(NH)C(NH2)2	116		+	++			C, CD	
–HCOOH, (NH)C(NH2)2	70		++				CD	80 %
C(NH2)3	60	+	+				CZ	
Asp	134		++	+			C, CG	
–H2O	116		++	+			C, CG	
–HCOOH	88	+	++				CG	
Cys							C	
Gly							C	
Glu	148		+	++			C, CD	
–H2O	130		+	++			C, CD	
–HCOOH	102		++				CD	
–HCOOH, H2O	84		++				CD	
His	156	+	++	+			CB, CG, CD2	
Ile	132		++	+			C, CD	
–HCOOH	86	+	++				CD	62 %
Leu	132		++				C	
–HCOOH	86	++					–	
Lys	147	+	++	+			C, CG	
–NH3	130	+	++	+			C, CG	
–NH3, HCOOH	84	+	++				CG	~60 %
Met	150		++	+			C, CG	
–NH3	133		++	+			C, CG	
–HCOOH	104	+	+				CG	63 %
Phe	166	+	+	++	+		C, CZ, Ce1, Ce2	
–HCOOH	120	+	++	+			CZ, Ce1, Ce2	
Pro	116		+	++			C, CD	
–HCOOH	70		++	+			CD	~99 %
Ser	106	+	++				C	
–H2O	88	+	++				C	
–HCOOH	60	++					–	
Thr	120		++	+			C, CG	
–H2O	102		++	+			C, CG	
–H2O, H2O	84		++	+			C, CG	
–HCOOH	74	+	++				CG	67 %
–HCOOH, H2O	56		++				CG	67 %
Trp	205	+	+	+	++	+	C, CG, CD1, CH2, Cz2, Cz3	
Tyr	182	+	+	++	+		C, CZ, Ce1, Ce2	
–NH3	165	+	+	++	+		C, CZ, Ce1, Ce2	
–H2O, H2O	146			++	+	+	C, CZ, Ce1, Ce2	
–HCOOH	136	+	++	+			CZ, Ce1, Ce2	
–HCOOH, NH3	119	+	++	+			CZ, Ce1, Ce2	
Val	118	+	++				C	
–HCOOH	72	++					–	

LC–ESI–MS/MS results (left) are compared with NMR results (right) and the labeling pattern expected from typical anabolic pathways. “++” indicates the most intense peak in the MS/MS spectra. Other peaks are indicated only if their intensities are at least 10 % of that of the most intense peak. Lower case letters in the atom labels indicate expectation of lighter labeling because symmetric carbons of 3-dehydroquinate can arise equally from labeled and unlabeled carbons of erythrose 4-phosphate in the shikimate pathway

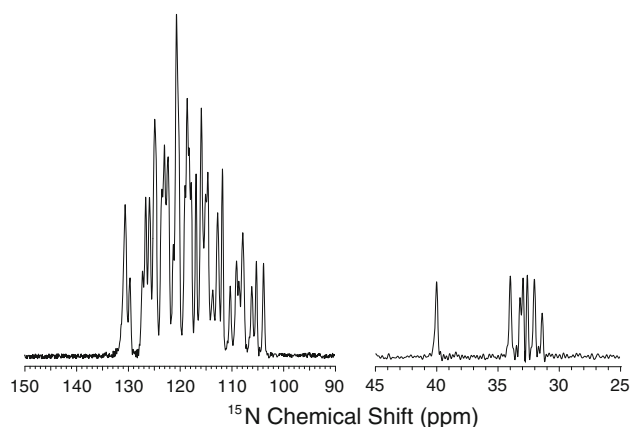
**Fig. 2** The MS spectrum of valine obtained by [redox- $^{13}\text{C}$ ] labeling compared, in the *inset*, to the natural abundance result



### Magic angle spinning NMR of microcrystalline GB1

The  $^{15}\text{N}$  spectrum of redox-labeled GB1 is shown in Fig. 3. The high resolution and the agreement of chemical shifts with previously published data, show that the redox-labeling protocol produced a homogenous sample consistent with previous studies (Franks et al. 2005).

The 1D  $^{13}\text{C}$  CP spectrum of redox-labeled microcrystalline GB1 is compared in Fig. 4 with the 1D  $^{13}\text{C}$  CP spectrum of uniformly labeled GB1. We see that the observed peaks are generally consistent with extensive carbonyl labeling and the sidechain labeling pattern shown in Fig. 1. In the carbonyl region, removal of the one-bond CA-CO scalar couplings has clearly improved the

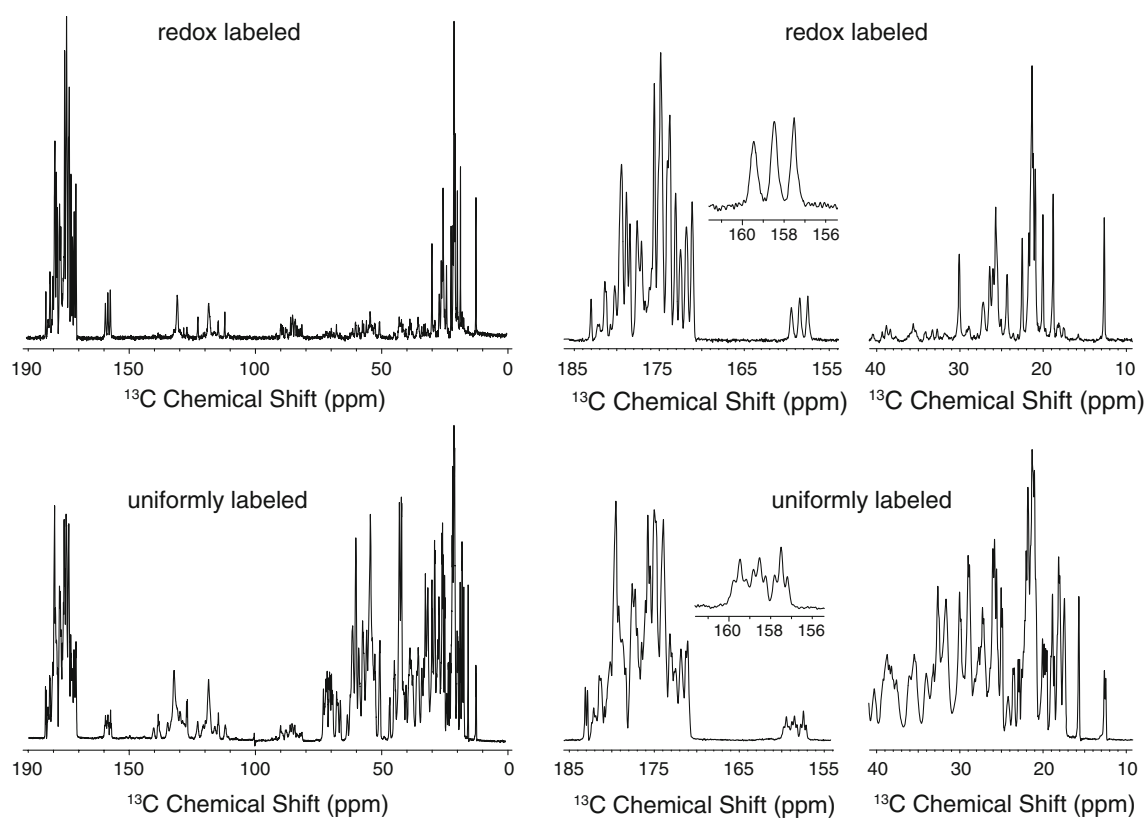


**Fig. 3**  $^{15}\text{N}$  CP MAS spectrum of [ $^{15}\text{N}$ , redox- $^{13}\text{C}$ ] labeled microcrystalline GB1. The spectrum was obtained at 800 MHz  $^1\text{H}$  Larmor frequency with 18.0 kHz MAS and 81 kHz TPPM  $^1\text{H}$  decoupling; 1,024 points were acquired with a 50  $\mu\text{s}$  dwell time for a total acquisition time of  $\sim 51$  ms

resolution such that over 20 distinct peaks are present. Also resolved to the baseline are the CZ signals of the three tyrosines from the primary sequence (between 156 and 160 ppm), although they are somewhat broadened at the base by light labeling at the neighboring Ce positions. At 12.7 ppm, the CD signal of Ile 6, which shows a splitting from one-bond scalar coupling in the uniformly labeled sample, is collapsed to a single, narrow peak in the redox-labeled sample. With the exception of several threonine CG contributions, almost all the aliphatic signals can be resolved in 1D. Since GB1 has no Pro, His, or Arg residues, we do not observe signals from those sidechains.

The one surprise was in the distribution of label in the tryptophan sidechain. CG and CD1, originating in ribose, were labeled as expected. However, the benzene ring, originating from chorismate, was more heavily labeled at CE2 than at CH2. In contrast, the pathways from chorismate to phenylalanine and tyrosine deposit aromatic labels in the pattern that is expected, indicating that there is nothing anomalous about the labeling of chorismate. Looking downstream from chorismate, it is as though more of the tryptophan derived from the p-aminobenzoate (PABA) product of chorismate than from the o-aminobenzoate (anthranilate) product of chorismate. However, we know of no precedent for such an alternative pathway and wonder whether it may be unique to *Pseudomonas mendocina* Palleroni.

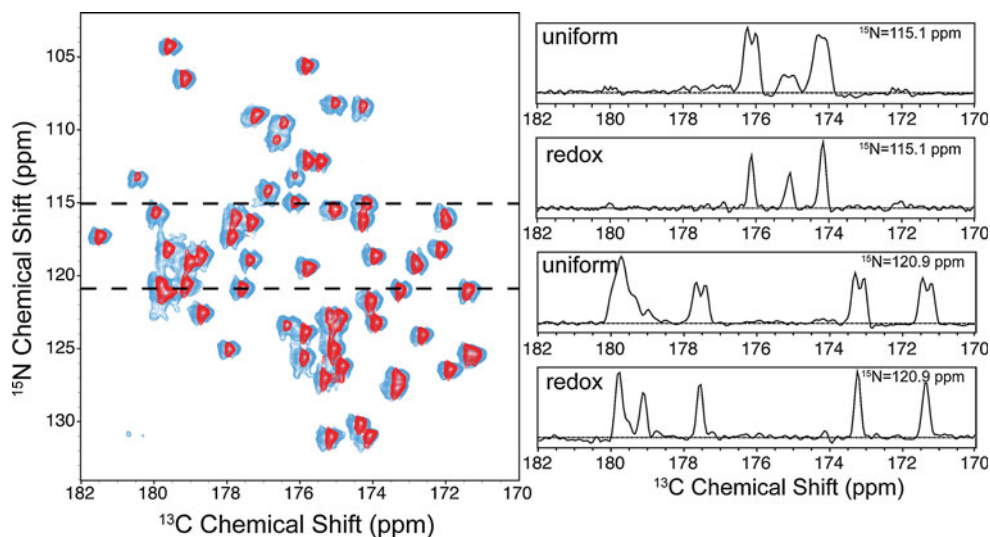
To further characterize the new labeling scheme we carried out a 2D  $^{15}\text{N}$ - $^{13}\text{C}$  ZF-TEDOR experiment with a mixing time optimized for maximum one-bond transfers ( $\sim 1.2$  ms TEDOR mixing). The ZF-TEDOR spectrum of redox-labeled GB1 is shown in Fig. 5. This spectrum shows that the carbonyl sites are all represented in redox-



**Fig. 4** 1D  $^{13}\text{C}$  CPMAS spectra of redox-labeled (*top*) compared with uniformly labeled (*bottom*) GB1. Full spectra are shown on the *left*, expansions of the carbonyl and aliphatic regions are shown on the *right*, and the tyrosine CZ peaks are shown in the *insets*. All three Tyr CZ sites are resolved for redox-labeled GB1. Both spectra were

acquired at 800 MHz  $^1\text{H}$ , 18 kHz MAS, with 83 kHz TPPM  $^1\text{H}$  decoupling and 128 scans with a 2.5 s recycle delay. Signals in the region 80–90 ppm are spinning sidebands. For the redox-labeled GB1, the small CA signals (30–70 ppm) are due to a combination of natural abundance background and low level scrambling

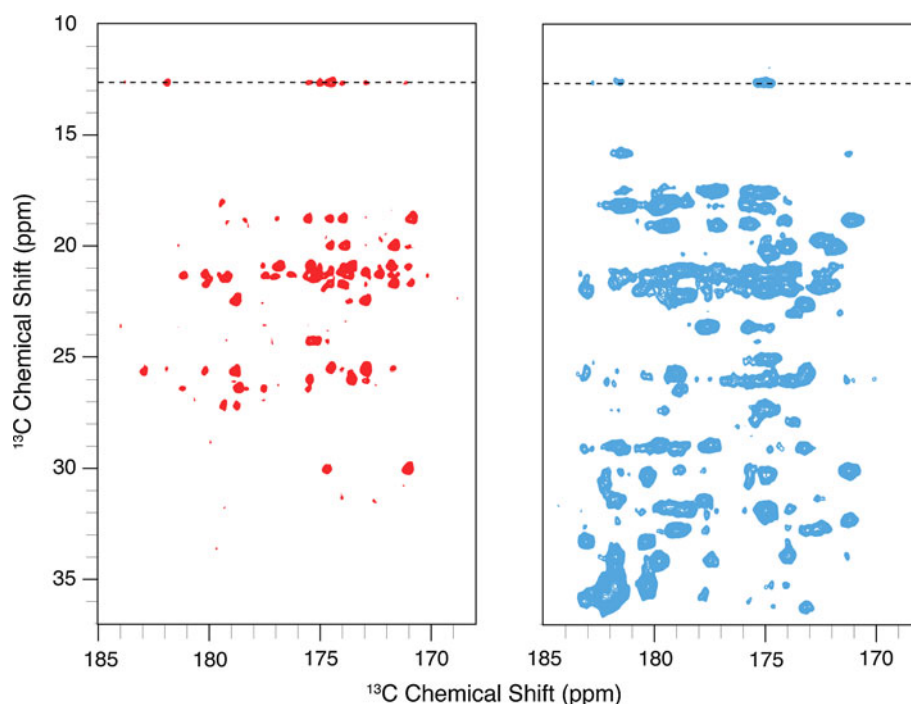
**Fig. 5** Comparison of 2D  $^{15}\text{N}$ - $^{13}\text{C}$  ZF-TEDOR spectra of uniformly labeled and redox-labeled GB1. *Left* spectrum for the redox-labeled sample (*dark red shade*) superimposed on the spectrum for the uniformly labeled sample (*light blue shade*). *Right* selected slices at  $^{15}\text{N}$  frequencies of 115.1 ppm (top two slices) and 120.9 ppm (bottom two slices). Spectra were recorded at 18.0 kHz MAS, 800 MHz  $^1\text{H}$  Larmor frequency. The full spectra are shown in Fig. S3



labeled GB1 and that removing the one-bond CA–CO scalar coupling reduces the CO linewidths by a factor of two or better. All peaks were assigned and consistent within 0.2–0.1 ppm of previously published shifts (Franks et al. 2005).

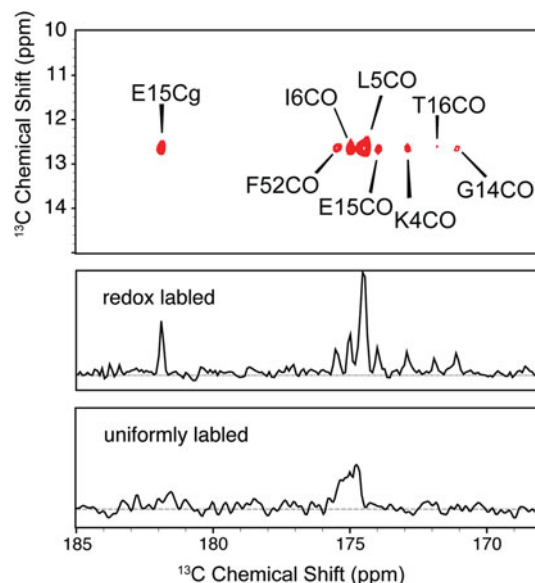
Figure 5 also shows the ZF-TEDOR spectrum of uniformly labeled GB1, acquired and processed with the same parameters. The comparison illustrates the reduction in linewidths and improvement in sensitivity obtained by redox-labeling. Although about half as much redox-labeled material

**Fig. 6** 12.4 ms RFDR  $^{13}\text{C}$ - $^{13}\text{C}$  2D spectra comparing redox-labeled GB1 (*left*) with uniformly labeled GB1 (*right*). These expansions show the CO-aliphatic region of the spectrum, which is typically very crowded for uniformly labeled samples. Each experiment was acquired at 18 kHz MAS with 8 scans per point for a total time of approximately 10 h per experiment. Both spectra were processed with identical parameters and displayed at the same contour level. Traces along the *dotted line* at 12.7 ppm are shown in Fig. 7. The full spectra are shown in Figs. S1 and S2



was present as uniformly-labeled material, the removal of all scalar couplings resulted in similar sensitivity.

To see if the redox labeling attenuated dipolar truncation, we acquired 2D  $^{13}\text{C}$ - $^{13}\text{C}$  RFDR correlation experiments on redox-labeled and uniformly labeled GB1 samples. The experiment employed 12.44 ms RFDR mixing, which is sufficient to generate long-range (5 Å or greater) transfers, at 18 kHz MAS and 800 MHz  $^1\text{H}$  Larmor frequency. Figure 6 shows an expansion of the CO-aliphatic region of the spectra, where we would expect to find intra- and inter-residue backbone CO to sidechain contacts with this amount of RFDR mixing. For uniformly labeled GB1, we see very few long range contacts; most contacts present are one and two-bond contacts with a few three bond or longer contacts. This is consistent with the observation that dipolar truncation would severely minimize transfers among weakly coupled spins. The panel for redox-labeled GB1 shows many intra- and inter-residue backbone to sidechain contacts, including contacts that are nearly 6 Å apart [according to PDB 2GI9 (Franks et al. 2006)]. Such contacts, illustrated in Fig. 7 by the portion of the RFDR spectrum involving the CD signal of Ile 6, provide experimental evidence that dipolar truncation is effectively avoided by the redox-labeling scheme. Furthermore, cross peak line widths for redox-labeled GB1 are significantly narrower, by a factor of 2 or more, compared with uniformly labeled GB1, permitting long-range contacts to be identified in larger systems despite the fact that the carbonyl region typically has a relatively small dispersion.



**Fig. 7** Details from Fig. 6 associated with the CD resonance of Ile 6 at 12.7 ppm. *Top panel* expansion of the RFDR spectrum for the redox-labeled sample, showing 8 unique cross peaks, including several corresponding to long-range, through space contacts. *Middle and bottom panels* 1D traces through 12.7 ppm for redox-labeled GB1 (*middle panel*) and uniformly labeled GB1 (*bottom panel*)

## Discussion

In this paper we have presented a new labeling scheme for isotopic labeling of *E. coli* recombinantly expressed proteins. The new scheme produces carbonyl labeling for all residue types except histidine and labeling of selected



sidechain sites for 12 residue types. The labeled peptone can be added to M9 minimal media and acts as a rich source of nutrients, enabling high protein yields.

This approach has a number of potential benefits and applications.

1. Foremost, as illustrated by the 2D RFDR data for GB1, dipolar truncation is effectively avoided. This will permit ZQ and DQ recoupling experiments to detect long-range contacts, and potentially measure long-range carbon–carbon distances with very high precision via a minimal set of broadband experiments. This use of the labeling scheme is being explored and a manuscript detailing this potential benefit will be forthcoming.
2. Redox labeling will also be useful for assignment experiments. In particular, it is suited to both 3D NCOCX experiments and 3D NCOCO experiments, analogous to experiments done by McDermott and coworkers (Goldbourn et al. 2007) on carbonyl labeled PF1 and Ladizhansky and coworkers on uniformly labeled GB1 (Janik et al. 2010). These experiments would significantly benefit from the removal of the scalar coupling contribution to linewidths; since this is achieved biochemically there would be no need for J-decoupling, which can be a source of polarization loss when implemented with selective pulses. Removal of scalar couplings also permits application of long mixing TEDOR experiments without the need for z-filters or selective pulses.
3. Redox labeling would improve measurements of order parameters. Backbone carbonyls (Dayie and Wagner 1997; Mulder and Akke 2003; Huang et al. 1999; Chang and Tjandra 2005) and methyl groups (Lee et al. 1999) have been very useful for site specific measurements of protein dynamics. Backbone CO and N order parameters have been also measured by TEDOR (Helmus et al. 2010). Such experiments would benefit from the removal of one bond scalar and dipolar couplings.
4. Salt bridges, which may contribute significantly to the stability of globular proteins (Anderson et al. 1990) and amyloid fibrils (Petkova et al. 2002), can be challenging to identify in uniformly labeled samples. Redox labeling highlights sidechain sites from several charged residues. This would facilitate measurement of salt bridge contacts.
5. Inverse redox labeling can be realized by starting with  $[2-^{13}\text{C}]$  acetate instead of  $[1-^{13}\text{C}]$  acetate. One potential use of such a scheme would be to acquire 3D NCACX experiments for intra-residue assignments. One bond CA-CO scalar couplings would be removed, thereby reducing linewidths for the CA

dimension in an NCACX spectrum. This would also improve measurements of CA order parameters.

6. The *Pseudomonas* peptone can support labeling in a variety of systems. In addition to the above described cultures of *E. coli* BL21(DE3) in M9 media, we have succeeded in using the peptone to grow *E. coli* MG1655 in modified LB media and *Saccharomyces cerevisiae* BY4742 in YPD media, with appropriate buffers.
7.  $[1-^{13}\text{C}]$  and  $[2-^{13}\text{C}]$  acetate also label *Pseudomonas* lipids and nucleic acids in the expected patterns. Thus, redox and inverse redox labeling may be useful for studies of biomolecules other than proteins.

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